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## cis-Dioxocyclam

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Molecules of 1,4,8,11-tetraazacyclotetradecane-5,7-dione, or cis-dioxocyclam, $\mathrm{C}_{10} \mathrm{H}_{20} \mathrm{~N}_{4} \mathrm{O}_{2}$, lie across mirror planes in space group Cmca; the crystal structure reveals interleaved columns of cis-dioxocyclam molecules along the $2_{1}$ screw axis parallel to the crystallographic $b$ axis. The columns are interconnected in a chain-like arrangement by an amido hydrogen-bonding network ( $\mathrm{N} \cdots \mathrm{O}=2.816 \AA$ ) and an amino hydrogen-bonding network $(\mathrm{N} \cdots \mathrm{N}=3.193 \AA)$. The intracolumn spacing is 9.02 Å.

## Comment

cis-Dioxocyclam, (I), is a diamido-diamino derivative of the well studied macrocycle 1,4,8,11-tetraazacyclotetradecane, or cyclam. Whereas both cyclam and cis-dioxocyclam contain four nitrogen donors for the chelation of transition metal ions, the latter provides the dual features of a macrocyclic polyamine and an oligopeptide (Kimura, 1986). Recently, cyclam was reported as a supramolecular synthon for the encapsulation of superanions (Airey et al., 1999). The crystal structure of cyclam consists of intramolecular hydrogen-bonded tetraamine macrocycles stacked in a columnar array via intermolecular hydrogen bonds (Airey et al., 1999). The present study addresses perturbation of the stacking resulting from changes in the hydrogen-bonding network upon substitution of two of the amines by amides. The crystal structure of transdioxocyclam has been reported previously as the dihydrate (Hubin et al., 2001; Frémond et al., 2000).

(I)

Compound (I) exists as a neutral molecule in the endodentate conformation (Subramanian \& Zaworotko, 1993). The structure of a single molecule of (I) is shown in Fig. 1. The amino H atom of N 1 is disordered, with $50 \%$ occupancy at position $\mathrm{H} 1 \mathrm{~N} A$, above the $\mathrm{N}_{4}$ plane of the molecule, and $50 \%$
occupancy at position $\mathrm{H} 1 \mathrm{~N} B$, directed toward the centre of the $\mathrm{N}_{4}$ plane. The resulting $\mathrm{N} 1 \cdots \mathrm{~N} 1^{\prime}$ distance of 2.967 (1) $\AA$ and $\mathrm{N} 1-\mathrm{H} 1 \mathrm{~N} B^{\prime}-\mathrm{N} 1^{\prime}$ angle of $131.0(14)^{\circ}$ are similar to the intramolecular hydrogen bonds reported for cyclam, with average values of $2.941 \AA$ and $134.4^{\circ}$, respectively.

An extensive intermolecular hydrogen-bonding network links each molecule of (I) to three neighbours (Fig. 2). The amido atoms N 2 and $\mathrm{N} 2^{\prime}$ serve as hydrogen-bond donors to the amido O atoms of a single neighbouring molecule. This pair of hydrogen bonds yields a 12 -membered ring, with $\mathrm{N} \cdots \mathrm{O}$ distances of 2.815 (1) $\AA$ and an $\mathrm{N}-\mathrm{H}-\mathrm{O}$ angle of $164.9(13)^{\circ}$. In addition, each molecule of (I) acts as a hydrogen-bond acceptor via O 1 and $\mathrm{O}^{\prime}$ in an identical interaction with an alternate neighbour. These four intermolecular hydrogen bonds result in an interleaved columnar array along the $2_{1}$ screw axis parallel to $b$.

The centroid-to-centroid distance between stacked molecules of (I) is long ( $9.02 \AA$ ), as expected. In addition, the amino moiety atoms N 1 and $\mathrm{N} 1^{\prime}$ are also involved in the intermolecular hydrogen bonding. The disordered positioning of atom H1N is interpreted as alternating the ordering of the H atom and the lone pair of electrons of each N atom, resulting in an eight-membered ring. This ring, comprising atoms $\mathrm{N} 1, \mathrm{~N} 1^{\prime}, \mathrm{H} 1 \mathrm{~N} A$ and $\mathrm{H} 1 \mathrm{~N} B^{\prime}$ on neighbouring molecules, is held together by covalent $\mathrm{N}-\mathrm{H}$ bonds, an intramolecular $\mathrm{N} 1 \cdots \mathrm{H} 1 \mathrm{~N} B^{\prime}-\mathrm{N} 1^{\prime}$ hydrogen bond and an intermolecular $\mathrm{N} 1-\mathrm{H} 1 \mathrm{~N} A \cdots \mathrm{~N} 1^{\prime}$ hydrogen bond. This last has an $\mathrm{N} \cdots \mathrm{N}$ distance of $3.190(1) \AA$ and an $\mathrm{N} 1-\mathrm{H} 1 \mathrm{~N} A \cdots \mathrm{~N} 1^{\prime}$ angle of $173.9(15)^{\circ}$.


Figure 1
A view of the molecule of (I) with the atom-numbering scheme. Displacement ellipsoids are shown at the $50 \%$ probability level and $H$ atoms are shown as small spheres of arbitrary radii.


Figure 2
A view of the hydrogen-bonding network in (I).

The amino hydrogen-bonding network further links the columnar array into a 'chain' of interleaved columns, with the cavities of the molecules of (I) aligned as parallel channels through the crystal (see supplementary figure). In total, the extended hydrogen-bonding network consists of one intramolecular and six intermolecular interactions per molecule of (I).

It is noteworthy that in the diprotonated form, as $\left[\mathrm{H}_{2}\right.$ dioxocyclam $]\left[\mathrm{PtCl}_{4}\right] \cdot \mathrm{H}_{2} \mathrm{O}$, the amido-O to amido-NH hydrogen-bonding network breaks down (Macíček et al., 1992). In $\left[\mathrm{H}_{2}\right.$ dioxocyclam $]\left[\mathrm{PtCl}_{4}\right] \cdot \mathrm{H}_{2} \mathrm{O}$, one amido O atom is hydrogen bonded to an amino- $\mathrm{NH}_{2}{ }^{+}$, while the other is hydrogen bonded to a water solvent molecule.

## Experimental

Compound (I) was synthesized from equimolar amounts of 3,7-diaza1,9 -nonanediamine and diethyl malonate under reflux conditions in ethanol ( $37.5 \mathrm{~m} M$ ) for 3 d , according to the published method of Tabushi et al. (1977). Recrystallization of the white solid from boiling ethanol yielded (I) as colourless blocks upon slow cooling for 15 h at 295 K.

## Crystal data

$\mathrm{C}_{10} \mathrm{H}_{20} \mathrm{~N}_{4} \mathrm{O}_{2}$
$M_{r}=228.30$
Orthorhombic, Cmca
$a=15.1714$ (9) £
$b=9.0190(5) \AA$
$c=17.2111(10) \AA$
$V=2355.0(2) \AA^{3}$
$Z=8$
$D_{x}=1.28 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}=1.23 \mathrm{Mg} \mathrm{m}^{-3}$

## Data collection

| Bruker SMART APEX CCD area- | 1455 independent reflections |
| :--- | :--- |
| $\quad$ detector diffractometer | 1359 reflections with $I>2 \sigma(I)$ |
| $\omega$ scans | $R_{\text {int }}=0.017$ |
| Absorption correction: multi-scan | $\theta_{\max }=28.3^{\circ}$ |
| $\quad(S A D A B S ;$ Sheldrick, 2001) | $h=-20 \rightarrow 19$ |
| $T_{\min }=0.96, T_{\max }=0.97$ | $k=-11 \rightarrow 11$ |
| 9566 measured reflections | $l=-22 \rightarrow 21$ |

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.034$
$w R\left(F^{2}\right)=0.088$
$S=1.08$
1455 reflections
98 parameters
H atoms treated by a mixture of independent and constrained refinement
$D_{m}$ measured by picnometry
Mo $K \alpha$ radiation
Cell parameters from 7156 reflections
$\theta=2.4-28.3^{\circ}$
$\mu=0.09 \mathrm{~mm}^{-1}$
$T=100$ (2) K
Block, colourless
$0.45 \times 0.37 \times 0.32 \mathrm{~mm}$

Table 2
Hydrogen-bonding geometry $\left(\AA{ }^{\circ}{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :---: | :---: | :---: | :---: |
| $\mathrm{~N} 1-\mathrm{H} 1 \mathrm{~N} A \cdots \mathrm{~N} 1^{\mathrm{i}}$ | $0.80(2)$ | $2.40(2)$ | $3.190(1)$ | $173.9(15)$ |
| $\mathrm{N} 1-\mathrm{H} 1 \mathrm{~N} B \cdots \mathrm{~N} 1^{\text {ii }}$ | $0.84(2)$ | $2.35(2)$ | $2.967(1)$ | $131.0(14)$ |
| $\mathrm{N} 2-\mathrm{H} 2 \mathrm{~N} A \cdots \mathrm{O} 1^{\text {iii }}$ | $0.87(1)$ | $1.97(2)$ | $2.815(1)$ | $164.9(13)$ |
| Symmetry codes: (i) $x, 1-y, 1-z ;$ (ii) $-x, y, z ;$ (iii) $x, \frac{1}{2}+y, \frac{1}{2}-z$. |  |  |  |  |

parameters for the methylene H atoms were allowed to refine, while their positions were constrained so that $\mathrm{C}-\mathrm{H}=0.99 \mathrm{~A}$.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXTL (Bruker, 2001).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: BK1641). An additional figure is also available. Services for accessing these data are described at the back of the journal.

## References

Airey, S., Drljaca, A., Hardie, M. J. \& Raston, C. L. (1999). J. Chem. Soc. Chem. Commun. pp. 1137-1138.
Bruker (2001). SAINT (Version 6.02a), SMART (Version 5.624) and SHELXTL (Version 6.12). Bruker AXS Inc., Madison, Wisconsin, USA.
Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Frémond, L., Espinosa, E., Meyer, M., Denat, F., Guilard, R., Huch, V. \& Veith, M. (2000). New J. Chem. 24, 959-966.
Hubin, T. J., Tyryshkin, N., Alcock, N. W. \& Busch, D. H. (2001). Acta Cryst. C57, 359-362.
Kimura, E. (1986). Coord. Chem. 15, 1-28.
Macíček, J., Gencheva, G., Mitewa, M., Bontchev, P. R., Lampeka, Y. \& Gavrish, S. (1992). J. Inclusion Phenom. Macrocycl. Chem. 13, 195-202.
Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
Sheldrick, G. M. (2001). SADABS. Version 2.03. University of Göttingen, Germany.
Subramanian, S. \& Zaworotko, M. J. (1993). J. Chem. Soc. Chem. Commun. pp. 952-954.
Tabushi, I., Taniguchi, Y. \& Kato, H. (1977). Tetrahedron Lett. 12, 10491052.

